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New sesquiterpenes from Inula japonica Thunb. with their inhibitory activities against LPS-induced NO production in RAW264.7 macrophages

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1. Introduction

Inula is a very important genus comprising about 100 species in the family Asteraceae.^{1,2} Plants belonging to this genus show high diversity in their secondary metabolites as well as in pharmaco-logical effects.^{[3](#page-9-0)} Inula japonica Thunb. is well known in China as 'Jinfeicao' and its aerial parts are used in traditional Chinese medicine for the treatment of various diseases such as tracheitis, bronchitis, hepatitis, and alimentary tract carcinoma.^{[1,2](#page-9-0)} Modern pharmacological studies have exhibited its diverse biological activities, such as anti-inflammatory, antifungal, antibacterial, antidiabetic, and hypolipidemic effects. $3-9$ $3-9$ $3-9$ In the previous studies, 12 dimeric sesquiterpenes and 4 diterpenes have been reported.^{[7](#page-9-0)-[9](#page-9-0)} As a part of our ongoing research program for bioactive secondary metabolites from Inula genus, the phytochemical analysis of I. japonica was further progressed and resulted the isolation and identification of 22 new sesquiterpenes $(1–22)$ together with 15 known ones $(23-37)$. In this paper, we described the isolation and structure elucidation of these new sesquiterpenes. Moreover, the inhibitory activities of all 37 isolates against LPS-induced NO production in RAW264.7 macrophages were also evaluated.

ABSTRACT

Twenty-two new sesquiterpenes were isolated from the aerial parts of Inula japonica Thunb., together with fifteen known ones. Their structures were determined by detailed spectroscopic analysis, X-ray diffraction studies, and modified Mosher method. All 37 compounds were evaluated for the inhibition of LPS-induced nitric oxide (NO) production in RAW264.7 macrophages, and most of isolates significantly inhibited the NO production with IC₅₀ values in the range of $3.5-20$ µM. Besides, results obtained in our studies provided a structure–activity relationship that would be used to design anti-inflammatory agents in the future.

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2. Results and discussion

2.1. Structure elucidation of new sesquiterpenes

The dried aerial parts of I. japonica were powdered and extracted with 95% ethanol and the extract was successively partitioned with petroleum ether, CH_2Cl_2 , EtOAc, and *n*-BuOH, respectively. The $CH₂Cl₂$ fraction was subjected to column chromatography over silica gel, Sephadex LH-20 and preparative HPLC to afford 22 new sesquiterpenes (1–22), together with 15 known ones: 5 α -hydrox-
vasperilin (23)¹⁰ 18-hydroxyalantolactone (24)¹¹ isoiyasperin yasperilin (23),^{[10](#page-9-0)} 1 β -hydroxyalantolactone (24),¹¹ isoivasperin
(25)¹² iyangustin (26)¹³ 1.6 dibydroxyeriolanolide (27)¹⁴ 1-20e (25) ,^{[12](#page-9-0)} ivangustin (26),¹³ 1,6 α -dihydroxyeriolanolide (27),^{[14](#page-9-0)} 1-acetoxy-6 α -hydroxyeriolanolide (28),¹⁴ 1 β -hydroxy-8 β -acetoxycostic acid methyl ester (29) , 6 1 β -hydroxy-8 β -acetoxy-isocostic acid methyl ester (30) , ^{[6](#page-9-0)} 4H-xanthalongin (31) , ^{[11](#page-9-0)} xanthalongin (32) , ¹⁵ eupatolide (33),^{[16](#page-9-0)} 7-epiloliolide (34),^{[17](#page-9-0)} vomifoliol (35),^{[18](#page-9-0)} corchoio-nol C (36),^{[19](#page-9-0)} grasshopper ketone (37)^{[20](#page-9-0)} ([Fig. 1](#page-1-0)).

Compound 1 was obtained as optically active, colorless bulk crystals. The molecular formula $C_{15}H_{22}O_4$, indicating five degrees of unsaturation, was established by HRESIMS (m/z 289.1411 for $[M+Na]^+$, calcd m/z 289.1416). The IR spectrum of 1 showed bands characteristic of hydroxyl (3489, 3354 cm $^{-1}$), carbonyl (1751 cm $^{-1}$), and olefinic bond (1664 cm^{-1}). These observations were in agreement with the observation of signals in the 13 C and DEPT NMR spectra ([Table 2\)](#page-4-0) for two oxygenated methines (δ _C 75.4, C-1; δ _C 79.8,

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C-8), one oxygenated quaternary carbon (δ _C 76.8, C-5), one olefinic group (δ _C 144.5, 120.9; C-11 and C-13), and one ester carbonyl (δ _C 173.2, C-12) accounting for two degrees of unsaturation. The remaining degrees of unsaturation were due to the presence of tricyclic nucleus in the molecule. Furthermore, the ¹H NMR spec-trum of 1 ([Table 1](#page-3-0)) indicated the presence of one methyl singlet (δ_H 1.02, s, Me-14), one methyl doublet (δ _H 1.01, d, J=7.0 Hz, Me-15), two oxymethines (δ _H 3.81, dd, J=11.8, 4.3 Hz, H-1 and δ _H 4.64, m, H-8), and two disubstituted olefinic protons (δ _H 6.05, s, H-13a and $\delta_{\rm H}$ 5.62, s, H-13b). In the 1 H $-^1$ H COSY experiments, the correlations of H-1 through H_2 -2, H_2 -3 and H-4 to H_3 -15, and H_2 -6 through H-7 and H-8 to H_2 -9 established two fragments (Fig. 2). The HMBC correlations traced from the methyls (H_3 -14 and H_3 -15) and olefinic proton $(H₂-13)$ suggested the presence of a eudesmane sesquiterpene moiety (Fig. 2). Some other key HMBC correlations between H-1/C-9, C-10 and Me-14; H-7/C-6, C-11, C-12 and C-13; and H-8/C-9 and C-10 were also observed. Moreover, the observed correlation of olefinic group (C-11 and 13), the ester carbon (C-12), and exocyclic olefinic protons (H-13a and 13b) authenticated the existence of a characteristic α -methylene lactone functionality. On the basis of above data, compound 1 was 1,5-dihydroxy-substituted eudesmane sesquiterpene lactone.

The relative stereochemistry of 1 was further confirmed by detailed analysis of NOESY spectra and an X-ray diffraction study ([Figs. 3 and 4](#page-2-0)). In the NOESY spectrum, the correlations of H-1/H-4, H3-14/H3-15, and H-7/H-8 were observed, which were in good agreement with the X-ray diffraction study. The absolute configuration was determined by modified Mosher method. $21,22$ The (S)- and (R) -MTPA esters of **1** (**1a** and **1b**, respectively) were prepared using the corresponding $(R)-(-)$ - and $(S)-(+)$ -MTPA chloride, respectively. The determination of $\Delta \delta$ values $(\delta_S - \delta_R)$ for protons
peigbboring C-1 led to the assignment of the R configuration at C-1 neighboring C-1 led to the assignment of the R configuration at C-1 in 1, while the $\Delta\delta$ value for methyl on C-4 was zero, possibly

Fig. 3.

because it lay on the MTPA plane (Fig. 5).^{[22](#page-9-0)} Therefore, all relevant chiral centers in 1 were assigned as 1R, 4S, 5R, 7R, 8R and 10S configurations on the basis of the $\Delta\delta$ results summarized in Fig. 3. Thus, 1 was elucidated as (1R,4S,5R,7R,8R,10S)-1,5-dihydroxyeudesma-11(13)-en-12,8-olide.

Compound 2 was shown to possess a molecular formula of $C_{15}H_{24}O_4$ (HRESIMS [M+Na]⁺, *m*/z 291.1544). The ¹H and ¹³C NMR data [\(Tables 1 and 2\)](#page-3-0) of 2 were comparable to those of 1 except for the absence of the signals assigned to the exocyclic olefinic protons H-13a (δ_H 6.05) and H-13b (δ_H 5.62) in 1 and the upfield shifts of the signals corresponding to the C-11 and C-13 protons from δ_C 144.5 and 120.9 in 1 to 42.3 and 9.9 in 2. Detailed assignments of the 1 H

and 13 C NMR data were determined on the basis of HSQC, 1 H $-^{1}$ H COSY, and HMBC experiments. The important NOESY correlations of H-11/H-7 and H-8 and other NOESY signals of 2 suggested the same relative configuration than 1. Therefore, 2 was identified as ¹b,5a-dihydroxy-4a,11aH-eudesma-12,8b-olide.

Compound 3 gave its molecular formula $C_{15}H_{22}O_4$, as established from HRESIMS (m/z 267.1587 for $[M+H]^+$). The ¹H NMR data ([Table 1\)](#page-3-0) of 3 were very similar to those of 5α -hydroxyasperilin (23) except for the absence of exocyclic olefinic protons H-13a (δ_H 6.07) and H-13b (δ ^H 5.67) in 23 and the presence of one methyl doublet at δ_H 1.16 in 3, which supported the hydrogenation of the double bond at C-11 and C-13 in 3. Detailed analysis of 2D NMR spectra further confirmed the planar structure and relative configuration of 3. The important NOESY correlations of H-11/H-7 and H-8 were observed. Thus, the structure of 3 was identified as 1β , 5 α -dihydroxy-11 α Heudesma-4(15)-en-12,8β-olide.

Both of compounds 4 and 5 possessed their molecular formula $C_{15}H_{20}O_4$, as shown from their positive HRESIMS at m/z 265.1446 $[M+H]^+$. The similarities between NMR spectra of 4 and 5 with known compound 1 β -hydroxyalantolactone (24) suggested the same skeleton.¹¹ Close comparison of 13 C NMR data of 4 and 24 indicated that 4 had an additional hydroxyl substituent at C-4, which was confirmed by analysis of 2D NMR spectra, including HSQC, 1 H $-{}^{1}$ H COSY, and HMBC. The relative configuration of 4 was established by NOESY experiment, and the significant correlations between H-6/H₃-15 and H-7 were observed. Thus, the final structure of 4 was elucidated as 1β , 4 β -dihydroxy-eudesma-5(6), 11(13)dien-12,8 β -olide. Similarly, comparison of the ¹H and ¹³C NMR data
of 5. (Tables 1, and 2), and **24** indicated that 5 has an additional of 5 ([Tables 1 and 2\)](#page-3-0) and 24 indicated that 5 has an additional hydroxyl group at C-3.^{[11](#page-9-0)} This was confirmed by the ${}^{1}H-{}^{1}H$ COSY

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Table 1
¹H NMR spectroscopic data for compounds **1–23** (J in Hz within parentheses)

Table 1 (continued)

	2 ^a	3 ^a	4 ^a	5 ^a	6 ^a
			2.45 m	2.28 m	
6.24 s; 5.67 s	6.27 s; 5.61 s	6.14 d (2.2) ; 5.74 d (2.0)	6.10 d (3.4) ; 5.55 d (3.1)	6.21 d (3.4) ; 5.50 d (3.1)	6.07 s; 5.67 s
1.02 s	1.02 s	1.26s	1.12 d (7.0)	1.14 d (7.0)	0.81 s
1.16s	1.21 s	1.11 $d(6.8)$	0.95s	1.02 s	4.86 s; 4.72 s
1.92s	1.95 s			2.06s	
3.73s	3.75s				
	No. 1^a				

^a Measured at 500 MHz in CD₃OD.
^b Measured at 500 MHz in CDCl₃.

a Measured at 125 MHz in CD₃OD.
^b Measured at 100 MHz in CDCl₃.
^c Measured at 125 MHz in CDCl₃.
^d Measured at 100 MHz in CDCl₃.

spectrum of the correlations from H-1 through H_2 -2, H-3, and H-4 to H3-15, as well as the long-range correlations of H-3 with C-1 and Me-15 in the HMBC spectrum. Based on the NOESY correlation of H-1/H-3, 5 was concluded to be 1β , 3 β -dihydroxy-4 α H-eudesma-5 (6) ,11(13)-dien-12,8 β -olide.

Both of 6 and 7 gave their molecular formula $C_{15}H_{20}O_4$ as established from their HRESIMS at m/z 263.1299 $[M-H]^-$, and m/z 265.1459 $[M+H]^+$, respectively. Their NMR data were very similar to those of known compound ivangustin (26) except for an

additional hydroxyl group at $C-3$.¹³ The main difference was observed in the NMR resonances mainly from C-1 through C-2, C-3 and C-4 to C-15 [\(Tables 1 and 2](#page-3-0)). In particular, the triplet of H-3 in 6 ($\delta_{\rm H}$ 4.08, t, J=8.4 Hz) was replaced by a doublet in 7 ($\delta_{\rm H}$ 4.01, d, J=4.5 Hz). Obviously, the upfield shift of C-1 in 7 (δ _C 68.5), in contrast to the corresponding shift value in **6** (δ _C 70.6), was due to a γ -gauche effect of α -OH at C-3.^{23–[26](#page-9-0)} The key NOESY correlation of H-1/H-3 was observed in **6** whereas the crucial NOESY correlations H-1/H-3 was observed in 6, whereas the crucial NOESY correlations of H-3/H-2 β (δ_H 1.97) and H₃-14/H-2 β (δ_H 1.97) were observed in 7. Thus, 6 was determined to be 1β , 3 β -dihydroxy-eudesma-4 (5) ,11(13)-dien-12,8 β -olide, and 7 was assigned as the C-3 epimer of 6 and named as 1β ,3 α -dihydroxy-eudesma-4(5),11(13)-dien- $12,8\beta$ -olide.

Compound 8 was assigned the molecular formula of $C_{15}H_{22}O_4$, as established from HRESIMS at m/z 267.1597 [M+H]⁺. Comparison of the NMR spectroscopic data of 8 with those of 7 showed these to be different in the characteristic α -methylene lactone functionality ([Tables 1 and 2\)](#page-3-0). The most significant features of NMR spectra of 8 were upfield shifted as exhibited by C-11, C-13, and H-7, and downfield shifted as exhibited by C-12, which indicated the absence of $\Delta^{11,13}$ exocyclic methylene group. In NOESY experiment, the crucial correlations of H-11/H-7 and H-8 were observed. On the basis of these data, 8 was concluded as 1β , 3 α -dihydroxy-11 α Heudesma-4(5)-en-12,8b-olide.

Compounds 9 and 10 had the molecular formula $C_{15}H_{18}O_4$ and $C_{15}H_{20}O_4$ as established from their HRESIMS m/z 263.1289 [M+H]⁺, and m/z 287.1239 [M+Na]⁺, respectively. The ¹H and ¹³C NMR spectra of 9 and 10 were very similar to those of 6 [\(Tables 1 and 2\)](#page-3-0), except that the hydroxymethine group at C-3 in 6 was replaced by a ketone carbonyl in 9, and one hydroxyl group at C-3 was replaced at C-2 in 10. Moreover, the important NOESY correlations of H-2/H-1, H-9 α /H-2, and H-8 in 10 showed that these two hydroxyl groups both have the β -configuration. Thus, **9** was elucidated as 1 β -hydroxy-3-oxo-eudesma-4(5),11(13)-dien-12,8 β -olide, and 10 was 1β ,2 β -dihydroxy-eudesma-4(5),11(13)-dien-12,8 β -olide.

Compound 11 was assigned the molecular formula of $C_{17}H_{24}O_5$, as established from its HRESIMS at m/z 331.1516 $[M+Na]^+$, accounting for six degrees of unsaturation. Absorption of hydroxyl (3537, 3350 cm⁻¹), carbonyl (1757, 1718 cm⁻¹), and olefinic bond (1656 cm $^{-1}$) was observed in its IR spectrum. A comparison of the ¹H and ¹³C NMR spectra of **11** with those of 1-acetoxy-6 α -hydrox-
veriolanolide (28) an isomer isolated from this study ¹⁴ Detailed yeriolanolide (28) , an isomer isolated from this study.^{[14](#page-9-0)} Detailed analysis of the 1D and 2D NMR spectra of 11 and 28 showed that they possessed the same skeleton except for the positional change of one acetoxyl group from C-1 for 28 to C-6 for 11 ([Fig. 2\)](#page-1-0). The HMBC correlation of H-6 to C-1' and the chemical shift of C-1 (δ_c) 62.7) confirmed that the acetoxyl group and a hydroxyl group were attached to C-6 and C-1 of 11, respectively.

The relative configuration of 11 was determined by NOESY experiment and coupling constants. The small coupling constant (1.5 Hz) between H-6 and H-7 implied a trans-configuration for these protons.¹⁴ Moreover, the allylic coupling observed between H-7 and H_2 -13 (2.5 and 2.0 Hz) suggested a cis-fused lactone ring of **11.**^{[27](#page-9-0)} The strong NOESY correlations of H-6/H₃-15 and H-7/H-8 were observed and gave a relative configuration of 11 [\(Fig. 3](#page-2-0)). Furthermore, a single crystal X-ray crystallographic measurement of 28 was also in agreement with the relative configuration of 11 [\(Fig. 4\)](#page-2-0). Thus, the structure of 11 was determined as 6α -acetoxy-1-hydroxy-⁴aH-1,10-secoeudesma-5(10),11(13)-dien-12,8b-olide.

Compound 12 exhibited a $[M+Na]^+$ ion peak at m/z 359.1832 in the positive HRESIMS, corresponding to the molecular formula, C₁₉H₂₈O₅. The ¹H and ¹³C NMR spectra of **12** were all comparable to those of 28 except for absence of an acetoxyl group attributed to C-1 of **28**, and the presence of an isobutyryl group [$\delta_{\rm H}$ 2.49 (m, H-2′), 1.14 (d, J=6.7 Hz, H-3'), and 1.16 (d, J=6.7 Hz, H-4'); δ_C 176.9 (C-1'), 34.1 (C-2'), 18.7 (C-3'), and 18.7 (C-4')] attributed to C-6 of **12** [\(Tables 1](#page-3-0) [and 2\)](#page-3-0). Thus, compound 12 was established as 6α -isobutyryloxy-1hydroxy-4aH-1,10-secoeudesma-5(10),11(13)-dien-12,8b-olide.

Both of 13 and 14 shared the same molecular formula $C_{20}H_{30}O_5$, and their ion peaks were at m/z 373.1987 [M+Na]⁺. Comparison of their 1D and 2D NMR data with those of 28 enabled the structure determination of both **13** and **14**. The $^1\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectroscopic data of 13 and 14 were very similar to those of 28 except for the presence of a 2-methybutyryl group [$\delta_{\rm H}$ 2.35 (m, H-2'), 1.51 $(m, H-3'a), 1.48$ $(m, H-3'b), 0.90$ $(t, J=7.0 Hz, H-4'),$ and 1.13

(d, J=7.0 Hz, H-5'); δ _C 176.6 (C-1'), 41.4 (C-2'), 26.5 (C-3'), 11.6 (C-4'), and 16.5 (C-5')] for **13** and an isovaleryl group $[\delta_H \, 2.17 \, (m, H-2'),$ 2.09 (m, H-3'), 0.96 (d, J=6.6 Hz, H-4'), and 0.96 (d, J=6.6 Hz, H-5'); δ _C 173.0 (C-1'), 43.6 (C-2'), 25.8 (C-3'), 22.4 (C-4'), and 22.4 (C-5')] for **14** at C-6 of them, instead of the acetoxyl group, which attached to C-1 of 28 [\(Tables 1 and 2](#page-3-0)). Therefore, the structures of compounds **13** and **14** were determined as 6α - $(2$ -methybutyryloxy)-1-hydroxy-4aH-1,10-secoeudesma-5(10),11(13)-dien-12,8bolide and 6a-isovaleryloxy-1-hydroxy-4aH-1,10-secoeudesma-5 (10) ,11 (13) -dien-12,8 β -olide, respectively.

Compound 15 gave a molecular formula $C_{21}H_{32}O_5$ from its HRESIMS at 365.2339 $[M+H]^+$, and exhibited very similar physical and spectroscopic data to those of 28 except for an additional 3-methyvaleryl group $[\delta_{\rm H}$ 2.32 (m, H-2'a), 2.11 (m, H-2'b), 1.87 $(m, H-3), 1.37$ $(m, H-4a), 1.26$ $(m, H-4b), 0.92$ $(t, J=7.5$ Hz, $H-5$ [']), and 0.94 (d, J=7.0 Hz, H-6'); δ_C 173.4 (C-1'), 41.3 (C-2'), 31.6 (C-3'), 28.8 (C-4'), 10.6 (C-5') and 18.7 (C-6')] for **15** at C-6, instead of the acetoxyl group, which attached to C-1 of 28 ([Tables 1 and 2\)](#page-3-0). Thus, compound 15 was elucidated as 6α -(3-methylvaleryloxy)-1-hydroxy-4aH-1,10-secoeudesma-5(10),11(13)-dien-12,8b-olide.

Compounds 16 and 17 were assigned the molecular formula $C_{15}H_{22}O_4$ and $C_{17}H_{24}O_6$ from their positive HRESIMS m/z 267.1604 $[M+H]^+$ and 325.1665 $[M+H]^+$, respectively. The ¹³C NMR spectroscopic data of 16 were similar to those of 1,6 α -dihydroxyeriolanolide (27) except that a methyl (δ_C 20.3, Me-14) and an oxygenated methine (δ _C 68.8, C-6) in 27 were replaced by an oxygenated methylene (δ_c 62.0, C-14) and a methylene (δ_c 30.0, C-6), respectively, in 16 ([Tables 1 and 2](#page-3-0)).¹⁴ Compound 16 was characterized as 1,14-dihydroxy-4aH-1,10-secoeudesma-5(10),11(13)-dien-12,8bolide. The NMR data of 17 compared with those of 28 showed that the only difference was the presence of an oxymethylene (δ_H 4.33 and 3.84; δ_C 62.1) in 17 instead of a methyl (δ_H 1.76, s; δ_C 20.3) in 28 ([Tables 1 and 2](#page-3-0)).¹⁴ Hence, **17** was elucidated as 1-acetoxy-6 α ,14dihydroxy-4 α H-1,10-secoeudesma-5(10),11(13)-dien-12,8 β -olide.

Compounds 18 and 19 had the same molecular formula $C_{18}H_{28}O_6$, established from their HRESIMS at m/z 341.1978 [M+H]⁺. Both of the NMR data of 18 and 19 showed a great similarity with those of 1 β -hydroxy-8 β -acetoxycostic acid methyl ester (29). In 18, one oxygenated quaternary carbon (δ _C 71.2, C-4) and a methyl (δ _H 1.20, s; δ_c 30.1) were appeared instead of the $\Delta^{4,15}$ exocyclic methylene group (δ_H 4.82, s and 4.61, s; δ_C 147.9 and 107.1) ([Tables 1](#page-3-0) and 2).^{[6](#page-9-0)} The similar relative configuration of **18** with **29** was deduced by a NOESY experiment, in which the key correlations of H₃-15/H-1 and H-7 were observed. Therefore, **18** was determined as ¹b,4b-dihydroxy-8b-acetoxy-5aH-eudesma-11(13)-en-12-oic acid methyl ester. Furthermore, the same planar structure of 19 as 18 was also found by analysis of 2D NMR spectra, including HSQC, $1H-1H$ COSY, and HMBC. By comparison of the related $1H$ and $13C$ NMR spectroscopic data of 19 with those of 18, 19 was found to be the C-4 epimer of 18. This was further confirmed by analysis of the NOESY correlations of 19, which revealed that H₃-14 was correlated with H₃-15. Thus, **19** was determined as 1β , 4 α -dihydroxy-8 β -acetoxy-5aH-eudesma-11(13)-en-12-oic acid methyl ester.

Compound 20 exhibited its molecular formula $C_{15}H_{22}O_4$, as deduced from its positive HRESIMS m/z 267.1594 [M+H]⁺, was the same as that of **27.**^{[14](#page-9-0)} A comparison of the 1 H NMR spectra of **20** with those of 27 revealed a great similarity except for the presence of a vinyl doublet (δ _H 5.25, d, J=4.2) in **20** and the absence of an oxymethines (δ_H 4.16, d, J=2.0) in 27 ([Table 1](#page-3-0)), implied the posi-tional changes of a double bond and a hydroxyl group.^{[14](#page-9-0)} This presumption above was confirmed by the 1 H $-{}^{1}$ H COSY correlations from H-6 through H-7 and H-8 to H_2 -9, as well as the key correlations of H₃-14 with C-5, C-9, and C-10 in the HMBC spectrum. Compared with 27, the similar NOESY correlations of 20 determined its relative stereochemistry. The NOESY correlations of H- $4/H₃$ -14 placed Me-14 at the α -configuration and OH-10 at the

 β -configuration. Thus, compound 20 was determined as 1,10 β dihydroxy-4 α H-1,10-secoeudesma-5(6),11(13)-dien-12,8 β -olide.

Compounds 21 and 22 had the molecular formula of $C_{15}H_{22}O_4$ and C₁₇H₂₄O₅ as established from their HRESIMS at m/z 267.1597 $[M+H]^+$ and m/z 309.1696 $[M+H]^+$, respectively. NMR spectroscopic data of 21 were very similar to those of 22, except for an additional acetoxyl group (δ ^c 171.2 and 21.1; δ _H 2.06) in **22**. The characteristic signals of 21 δ _H 6.10 (d, J=3.4, H-13a) and 5.55 $(d, J=3.1, H-13b)$; δ_C 142.8 (C-11), 172.4 (C-12), and 120.4 (C-13)] in the 1 H and 13 C NMR ([Tables 1 and 2](#page-3-0)) along with an IR absorption bands at 1660 cm⁻¹ and 1739 cm⁻¹ indicated the existence of an α -
methylene, lactone, functionality, A, long-range, spin-system, of methylene lactone functionality. A long-range spin-system of CH2CHCHCH2CH (CH3)CHCHCH2CH [C-6/C-7/C-8/C-9/C-10 (C-14)/ C-1/C-2/C-3/C-4], combined with the significant HMBC correlations of H₃-14 to C-1, 9, and 10, H₃-15 to C-4, 5, and 6, and H₂-13 to C-7, 11, and 12, indicated the presence of a pseudoguaianolide moiety ([Fig. 2\)](#page-1-0). In addition, the chemical shifts of C-2 (δ_c 71.3) and C-4 (δ_c 79.3) implied that two hydroxyl groups were attached to C-2 and C-4 of the pseudoguaianolide moiety, respectively. Hence, the planar structure of 21 was constructed as 2,4-dihydroxy-pseudoguaianolide. The relative stereochemistry of 21 was mainly deduced from NOESY correlations of H-4/H-6 α , H-7/H-1, and H-6 α , and H₃-14/H-2, H-8, and H₃-15 [\(Fig. 3\)](#page-2-0). In the bargain, the large coupling constant (9.0 Hz) between H-7 and H-8 and the coupling constants between H-7 and H_2 -13 (3.4 and 3.1 Hz) further confirmed the trans-fused lactone ring.²⁷ Thus, compound 21 was elucidated as 2α , 4 β -dihydroxy-1a^H ¹⁰aH-pseudoguai-11(13)-en-12,8a-olide. Furthermore, The HMBC correlation between $\delta_{\rm H}$ 5.06 (H-2) and $\delta_{\rm C}$ 171.2 (C-1′) established the connection of the acetoxyl group to C-2 in 22 ([Tables 1 and 2\)](#page-3-0). Compound 22 was then concluded as 2α -acetoxy-⁴b-hydroxy-1aH,10aH-pseudoguai-11(13)-en-12,8a-olide.

2.2. Assay for inhibitory activities against NO production

As one of the largest groups of secondary plant metabolites, sesquiterpene lactones were reported to be the active components of many medicinal plants from the Asteraceae family and showed various biological activities such as anti-inflammatory, antiproliferative, and bactericidal effects. $28-30$ $28-30$ $28-30$ In particular, their potent anti-inflammatory property has received considerable attention and been reported to be mediated chemically by α , β -unsaturated carbonyl structures, such as an α -methylene- γ lactone or an α , β -unsubstituted cyclopentenone.^{[31](#page-9-0)} Therefore, it was meaningful to investigate the anti-inflammatory effects for these sesquiterpenes isolated from I. japonica. In this study, all 37 compounds were tested on their cytotoxic activities on RAW264.7 macrophages and showed no toxic at the dose evaluated (50 μ M), and then tested for inhibitory activities against LPS-induced NO production in this cell line under the concentration range from 1 to 50 μ M. The IC₅₀ values obtained suggested that most of these compounds significantly inhibited the NO production with IC_{50} values in the range of 3.5–20 μ M (Table 3) except compounds $2, 3, 8$, and $34-37$, which was attributed to the absence of α -methylene- γ -lactone. Interestingly, the lactone rings of compounds 18, 19, 29, 30 were broken; nevertheless they exhibited their IC_{50} values under the concentration of 20μ M. These unforeseen results were supposed to arise from the common propenoic methyl ester chain of these four sesquiterpenes. On the other hand, compound 24 showed stronger inhibitory effect than compounds 4, 5, and 25 because of the presence of an additional hydroxyl group at C-4, C-3, and C-2 in the latter ones, respectively, reduced cellular penetration of the compounds across the phospholipid bilayers surrounding the cells, and consequently decreased the anti-inflammatory activity[.32](#page-9-0) Similarly, the occurrence of an additional hydroxyl group at C-2 or C-3 (compounds 6, 7, and 10) clearly reduced the

Table 3

Inhibitory effects of compounds isolated from I. japonica against LPS-induced NO production in RAW264.7 macrophages $(n=4)$

Compounds	IC_{50}^a (μ M)	Compounds	IC_{50}^a (μ M)
1	7.1	20	7.3
$\overline{\mathbf{c}}$	20.3	21	9.6
3	20.5	22	3.5
4	8.7	23	9.2
5	6.0	24	5.1
6	8.8	25	12.7
7	9.0	26	5.0
8	22.1	27	18.6
9	7.3	28	10.9
10	12.6	29	18.6
11	8.1	30	17.4
12	8.8	31	6.9
13	4.8	32	18.3
14	4.8	33	3.5
15	4.3	34	49.7
16	17.2	35	25.0
17	15.7	36	22.6
18	18.9	37	33.5
19	13.1	AG ^b	0.6

 a Inhibitory effects of compounds $1-37$ against LPS-induced NO production in RAW264.7 macrophages.

 b Positve control (\geq 98.0%, Sigma); AG: aminoguanidine.</sup>

inhibitory effects on NO production compared to compounds 26 in which non-hydroxyl groups appeared at both C-2 and C-3. Moreover, the acylation of compound 27 at C-1 or C-6 (compounds 11-15 and 28) obviously augmented its activity and compound 15 exhibited the strongest activity with IC_{50} value of 4.3 µM due to its longest lipophilic chain 3-methyvaleryl group at C-6, which further verified the hypothesis above. In conclusion, the α -methylene- γ -lactone and the propenoic methyl ester chain were proposed to be the key chemical characteristic responsible for the above mentioned activities, and the lipophilicity of these compounds also was an important factor for their potential antiinflammatory activities.

3. Conclusion

In summary, we have fully described the isolation and structure elucidation of 22 new sesquiterpene derivatives and 15 known ones from the aerial parts of I. japonica. Sesquiterpenes show various interesting biological activities, including anti-inflammatory activities, $28-30$ $28-30$ $28-30$ therefore, the inhibitory activities of all 37 isolates on LPS-induced NO production in RAW264.7 macrophages were also evaluated. The obtained IC_{50} values demonstrated significant inhibitory activities of most of sesquiterpenes for NO production and a structure-activity relationship analysis had been discussed. These findings would provide information for the future design of anti-inflammatory agents.

4. Experimental

4.1. General procedures

Optical rotations were obtained with a JASCO P-2000 polarimeter. IR spectra were obtained with a Bruker FTIR Vector 22 spectrometer. 1D and 2D NMR spectra were recorded on a Bruker Avance-400 or Avance-500 spectrometers in CDCl₃ or CD₃OD with TMS as internal standard. ESIMS spectra were recorded on an Agilent LC/MSD Trap XCT spectrometer (Waters, USA), and HRE-SIMS on a Q-TOF micro mass spectrometer (Waters, USA). A preparative column (Shimadzu PRC-ODS EV0233) was used for preparative HPLC (Shimadzu LC-6AD). TLC analysis was run on HSGF₂₅₄ silica gel plates (10-40 µm, Yantai, China). Column chromatography was performed on silica gel (100-200, 200-300 mesh,

Yantai, China), silica gel H $(10-40 \,\mu m, Qingdao, China)$, and Sephadex LH-20 (Pharmacia Co. Ltd.).

4.2. Plant material

The aerial parts of I. japonica were collected in Anhui province, PR China, in October, 2006, and were authenticated by Professor Bao Kang Huang, Department of Pharmacognosy, School of Pharmacy, Second Military Medical University. A voucher specimen (No. 2007XFH1) was deposited at School of Pharmacy, Shanghai Jiao Tong University.

4.3. Extraction and isolation

The dried aerial parts of I. japonica (20.0 kg) were powdered and extracted with 95% ethanol (3×10 L) for three times (48 h, 24 h, and 24 h) at room temperature. The ethanolic extract was successively partitioned with petroleum ether (30 L), $CH₂Cl₂$ (40 L), EtOAc (30 L), and n-BuOH (30 L), respectively. The $CH₂Cl₂$ fraction (84.5 g) was chromatographed on a silica gel column eluting with a step gradient of CH2Cl2/MeOH (100:0, 50:1, 20:1, 10:1, 5:1, 2:1, 1:1) to give 11 fractions (Fr1–Fr11). Fr1 (5.6 g) and Fr2 (8.4 g) were combined and subjected to CC over macroporous resin MCI, Sephadex LH-20, and silica gel to give 24 (820.0 mg), 26 (311.1 mg), 28 (708.5 mg), 32 (15.0 mg) , and 33 (23.8 mg). Fr3 (16.3 g) was subjected to a silica gel CC with mixtures of PE/EtOAc (20:1, 10:1, 5:1, 2:1, 1:1, EtOAc) as eluents in a stepwise gradient mode to obtain nine fractions (Fr3-1-Fr3-9). Compounds 12 (27.0 mg), 13 (73.0 mg), 14 (73.0 mg), 15 (224.7 mg), 29 (7.0 mg), and 30 (26.9 mg) were isolated after CC over macroporous resin MCI followed by preparative HPLC (CH₃CN/ H2O, 50:50) from subfraction Fr3-4. From subfraction Fr3-8, compounds 11 (67.0 mg) and 34 (7.0 mg) were obtained after CC over Sephadex LH-20 (MeOH/CH₂Cl₂, 1:1) and preparative HPLC (CH₃CN/ H₂O, 35:65). By the same procedures, compound 22 (6.7 mg) was obtained from subfraction Fr3-9. Fr4 (5.2 g) was subjected to silica gel CC eluted with PE/EtOAc (15:1, 10:1, 5:1, 2:1, 1:1, EtOAc) to give seven fractions (Fr4-1–Fr4-7). Subfraction Fr4-1 was subjected to CC over Sephadex LH-20 (MeOH) and followed by preparative HPLC (MeOH/H₂O, 40:60) led to the isolation of **27** (204.5 mg) and **31** (16.7 mg). Similarly, 9 (9.0 mg), 18 (8.0 mg), 23 (20.1 mg), 35 (11.2 mg) , and 36 (5.9 mg) were obtained from Fr4-2, while 1 (91.5 mg) , $2(4.6 \text{ mg})$, $3(3.8 \text{ mg})$, $4(1.1 \text{ mg})$, $10(4.6 \text{ mg})$, $20(9.9 \text{ mg})$, **25** (4.0 mg), and **37** (3.4 mg) from Fr4-3. Fr6 (3.5 g) was subjected to CC over macroporous resin MCI, Sephadex LH-20 (MeOH), and preparative HPLC (MeOH/H₂O, 35:65) to give **5** (226.0 mg), **6** (4.7 mg), 16 (1.0 mg), and 17 (22.5 mg). By the same procedures, 19 (90.4 mg) was isolated from Fr5 (1.3 g), while 7 (36.8 mg), **8** (11.4 mg) , and 21 (11.3 mg) from Fr7 (3.2 g) . The purities of these compounds were ranging from 95.5 to 99.8% determined by HPLC.

4.3.1. Compound 1. Colorless bulk crystals; $[\alpha]_0^{20}$ +153.3 (c 0.10, 1060 H) and α and α and α and α and α and α in α and α in α and α in α and α in α in α in α in α in MeOH); IR (KBr) ν_{max} 3489, 3354, 2936, 1751, 1664, 1269, 1165, 1011, 967, 947 cm $^{-1}$; for 1 H and 13 C NMR data, see [Tables 1 and 2](#page-3-0); ESIMS (positive) m/z 289 [M+Na]⁺, 555 [2M+Na]⁺; ESIMS (negative) m/z 265 $[M-H]^-$, 531 $[2M-H]^-$; HRESIMS (positive) $[M+Na]^+$ m/z 289.1411 (calcd for C₁₅H₂₂O₄Na, 289.1416).

4.3.2. Compound 2. Amorphous powder; [α] $_0^{20}$ +21.8 (c 0.13, MeOH);
IR (KRr) $v = 3.461, 2933, 1749, 1410, 1374, 1770, 1169, 1025, 979$ cm⁻¹. IR (KBr) $\nu_{\rm max}$ 3461, 2933, 1749, 1410, 1374, 1270, 1169, 1025, 979 cm $^{-1};$ for ¹H and ¹³C NMR data, see [Tables 1 and 2](#page-3-0); ESIMS (positive) m/z 291 $[M+Na]^+$; ESIMS (negative) m/z 267 $[M-H]^-$; HRESIMS (positive) $[M+Na]^+$ m/z 291.1544 (calcd for C₁₅H₂₄O₄Na, 291.1567).

4.3.3. Compound 3. Amorphous powder; $[\alpha]_0^{20}$ +67.7 (c 0.12, MeQH): IR (KBr) $v = 3484, 2942, 2862, 1754, 1662, 1458, 1262$ MeOH); IR (KBr) $\nu_{\rm max}$ 3484, 2942, 2862, 1754, 1662, 1458, 1262, 1145, 1037, 1001 cm^{-1} ; for ¹H and ¹³C NMR data, see [Tables 1 and 2](#page-3-0); ESIMS (positive) m/z 267 [M+H]⁺; ESIMS (negative) m/z 265 $[M-H]^-$; HRESIMS (positive) $[M+H]^+$ m/z 267.1587 (calcd for C₁₅H₂₃O₄, 267.1596).

4.3.4. Compound **4**. Amorphous powder; [α] $_{D}^{20}$ +315.1 (c 0.03, MeQH) IR (KBr) $_{N}$ = 3461 3315 2933 1749 1662 1410 1337 1270 MeOH); IR (KBr) ν_{max} 3461, 3315, 2933, 1749, 1662, 1410, 1337, 1270, 1169, 1025, 979 cm^{-1} ; for ¹H and ¹³C NMR data, see [Tables 1 and 2](#page-3-0); ESIMS (positive) m/z 551 [2M+Na]⁺; ESIMS (negative) m/z 263 $[M-H]^-$, 527 [2 M–H]⁻; HRESIMS (positive) $[M+H]^+$ m/z 265.1446 (calcd for $C_{15}H_{21}O_4$, 265.1440).

4.3.5. Compound **5.** Amorphous powder; $[\alpha]_D^{20}$ +152.8 (c 0.11, MeQH): IR (KBr) α 3552 3240 2943 2554 2552 3552 1758 MeOH); IR (KBr) v_{max} 3552, 3240, 2943, 2614, 2568, 2362, 1758, 1660, 1466, 1340, 1266, 1155, 1034, 974 cm⁻¹; for ¹H and ¹³C NMR data, see [Tables 1 and 2;](#page-3-0) ESIMS (positive) m/z 287 [M+Na]⁺; ESIMS (negative) m/z 263 $[M-H]^-$, 527 $[2\ M-H]^-$; HRESIMS (positive) $[M+H]^+$ m/z 265.1446 (calcd for C₁₅H₂₁O₄, 265.1440).

4.3.6. Compound **6.** Amorphous powder; $[\alpha]_0^{20}$ +36.4 (c 0.10, MeQH): IR(KBr)_N 3331 2973 2927 2881 1924 1758 1662 1453 MeOH); IR (KBr) v_{max} 3331, 2973, 2927, 2881, 1924, 1758, 1662, 1453, 1420, 1379, 1088, 1046, 880, 804 cm⁻¹; for ¹H and ¹³C NMR data, see [Tables 1 and 2](#page-3-0); ESIMS (positive) m/z 287 [M+Na]⁺, 551 [2M+Na]⁺; ESIMS (negative) m/z 263 [M-H]⁻, 527 [2 M-H]⁻; HRESIMS (negative) $[M-H]$ [–] m/z 263.1299 (calcd for C₁₅H₁₉O₄, 263.1283).

4.3.7. Compound 7. Amorphous powder; $[\alpha]_0^{20}$ +80.6 (c 0.25, MeQH) IR (KBr) π 3330 2957 2882 2550 2257 1925 MeOH); IR (KBr) v_{max} 3330, 2973, 2882, 2545, 2350, 2257, 1925, 1753, 1663, 1452, 1379, 1088, 1046, 880 cm⁻¹; for ¹H and ¹³C NMR data, see [Tables 1 and 2](#page-3-0); ESIMS (positive) m/z 287 [M+Na]⁺, 551 $[2M+Na]^+$; ESIMS (negative) m/z 263 $[M-H]^-$; HRESIMS (positive) $[M+H]^+$ m/z 265.1459 (calcd for C₁₅H₂₁O₄, 265.1440).

4.3.8. Compound 8. Amorphous powder; $[\alpha]_D^{20}$ +205.4 (c 0.10, MeQH) IR (KBr) α 3322 2960 2530 2257 1025 1760 1652 1421 MeOH); IR (KBr) v_{max} 3322, 2960, 2530, 2257, 1925, 1760, 1652, 1421, 1329, 1088, 1046, 880 cm⁻¹; for ¹H and ¹³C NMR data, see [Tables 1](#page-3-0) [and 2;](#page-3-0) ESIMS (positive) m/z 289 [M+Na]⁺; ESIMS (negative) m/z 265 $[M-H]^-$; HRESIMS (positive) $[M+H]^+$ m/z 267.1597 (calcd for $C_{15}H_{23}O_4$, 267,1596).

4.3.9. Compound **9.** Amorphous powder; $[\alpha]_0^{20}$ +20.6 (c 0.20, MeQH): IR (KBr) $v = 3484, 2942, 2862, 2571, 1754, 1662, 1458$ MeOH); IR (KBr) v_{max} 3484, 2942, 2862, 2571, 1754, 1662, 1458, 1397, 1322, 1262, 1145, 1001, 812 cm⁻¹; for ¹H and ¹³C NMR data, see [Tables 1 and 2;](#page-3-0) ESIMS (positive) m/z 285 $[M+Na]^+$; HRESIMS (positive) $[M+H]^+$ m/z 263.1289 (calcd for C₁₅H₁₉O₄, 263.1283).

4.3.10. Compound **10**. Amorphous powder; $\left[\alpha\right]_0^{20}$ +90.6 (c 0.10, 0.10)
MeOH): IR (KBr) $v = 3317, 2973, 2926, 2881, 1925, 1754, 1658, 1454$ MeOH); IR (KBr) v_{max} 3317, 2973, 2926, 2881, 1925, 1754, 1658, 1454, $1420, 1379, 1328, 1274, 1088, 1046, 880, 803$ cm⁻¹; for ¹H and ¹³C NMR data, see [Tables 1 and 2;](#page-3-0) ESIMS (positive) m/z 287 [M+Na]⁺; HRESIMS (positive) $[M+Na]^+$ m/z 287.1239 (calcd for C₁₅H₂₀O₄Na, 287.1259).

4.3.11. Compound **11.** Amorphous powder; $[\alpha]_0^{20}$ –22.3 (c 0.22, GH_CU₂): IR (KBr) $v = 3537, 3350, 2932, 2866, 2133, 1757, 1718$ CH₂Cl₂); IR (KBr) v_{max} 3537, 3350, 2932, 2866, 2133, 1757, 1718, 1656, 1412, 1376, 1277, 1250, 1150, 1019, 979 cm⁻¹; for ¹H and ¹³C NMR data, see [Tables 1 and 2](#page-3-0); ESIMS (positive) m/z 331 [M+Na]⁺; ESIMS (negative) m/z 307 [M-H]⁻; HRESIMS (positive) [M+Na]⁺ m/z 331.1516 (calcd for C₁₇H₂₄O₅Na, 331.1521).

4.3.12. Compound **12.** Amorphous powder; $\left[\alpha\right]_0^{20}$ -28.0 (c 0.21, $\left[\alpha\right]$ CH₂CH₂ (kB_C) μ 3534 2930 1757 1720 1660 1375 1321 1248 CH₂Cl₂); IR (KBr) v_{max} 3534, 2930, 1757, 1720, 1660, 1375, 1321, 1248, 1043, 983 cm^{-1} ; for ¹H and ¹³C NMR data, see [Tables 1 and 2;](#page-3-0) ESIMS (positive) m/z 359 $[M+Na]^+$; HRESIMS (positive) $[M+Na]^+$ m/z 359.1832 (calcd for C19H28O5Na, 359.1834).

4.3.13. Compound **13.** Amorphous powder; $\left[\alpha\right]_0^{20}$ –20.9 (c 0.33, GH-CL): IR (KBr) $v = 3537, 3352, 2932, 2866, 2133, 1757, 1718, 1657$ CH_2Cl_2); IR (KBr) ν_{max} 3537, 3352, 2932, 2866, 2133, 1757, 1718, 1657,

1376, 1277, 1250, 1151, 1069, 1019, 980, 963 cm⁻¹; for ¹H and ¹³C NMR data, see [Tables 1 and 2;](#page-3-0) ESIMS (positive) m/z 373 [M+Na]⁺; HRESIMS (positive) $[M+Na]^+$ m/z 373.1987 (calcd for C₂₀H₃₀O₅Na, 373.1991).

4.3.14. Compound **14**. Amorphous powder; $\left[\alpha\right]_0^{20} - 21.1$ (c 0.33, $\left[\text{CH}_2(\text{L}) : \text{IR} \left(\text{KRr}\right)\right]$ 2540 2933 2860 1760 1720 1652 1375 1340 $CH₂Cl₂$); IR (KBr) ν_{max} 3540, 2933, 2860, 1760, 1720, 1652, 1375, 1340, 1277, 1250, 1129, 1020, 979, 923 cm⁻¹; for ¹H and ¹³C NMR data, see [Tables 1 and 2](#page-3-0); ESIMS (positive) m/z 373 [M+Na]⁺; HRESIMS (positive) $[M+Na]^+$ m/z 373.1897 (calcd for C₂₀H₃₀O₅Na, 373.1991).

4.3.15. Compound **15**. Amorphous powder; $\left[\alpha\right]_0^{20}$ –14.4 (c 0.13, $\left(113\right)$ CH-Cl-)[,] IR (KBr) $v = 3450$ 2935 1755 1729 1660 1377 1229 1129 $CH₂Cl₂$); IR (KBr) ν_{max} 3450, 2935, 1755, 1729, 1660, 1377, 1229, 1129, 980, 920 cm $^{-1}$; for 1 H and 13 C NMR data, see [Tables 1 and 2](#page-3-0); ESIMS (positive) m/z 387 $[M+Na]^+$, 751 $[M+Na]^+$; HRESIMS (positive) $[M+H]^+$ m/z 365.2339 (calcd for C₂₁H₃₃O₅, 365.2328).

4.3.16. Compound **16**. Amorphous powder; $[\alpha]_D^{20}$ +136.3 (c 0.02, MeQH): IR (KRr) v 3523, 3090, 2958, 2959, 2869, 1727, 1654 MeOH); IR (KBr) v_{max} 3523, 3090, 2958, 2925, 2869, 1727, 1654, 1418, 1357, 1282, 1073, 1031, 973 $\rm cm^{-1}$; for ¹H and ¹³C NMR data, see [Tables 1 and 2;](#page-3-0) ESIMS (positive) m/z 289 $[M+Na]^+$; HRESIMS (positive) $[M+H]^+$ m/z 267.1604 (calcd for C₁₅H₂₃O₄, 267.1596).

4.3.17. Compound **17.** Amorphous powder; [α]<mark>20</mark> +61.8 (c 0.15,
MeOH):IR(KBr)» - 3494 2937 2552 2361 1736 1654 1457 1406 MeOH); IR (KBr) v_{max} 3494, 2937, 2552, 2361, 1736, 1654, 1457, 1406, 1364, 1260, 1160, 1031, 952 cm⁻¹; for ¹H and ¹³C NMR data, see [Tables 1 and 2;](#page-3-0) ESIMS (positive) m/z 347 $[M+Na]^+$; HRESIMS (positive) $[M+H]^+$ m/z 325.1665 (calcd for C₁₇H₂₅O₆, 325.1651).

4.3.18. Compound **18**. Amorphous powder; $\left[\alpha\right]_0^{20}$ -18.4 (c 0.22, CH_2Cl_2) IR(KBr)_N 3519 3423 3361 2858 2638 2595 2362 1716 $CH₂Cl₂$; IR(KBr) ν_{max} 3519, 3423, 3361, 2858, 2638, 2595, 2362, 1716, 1669, 1458, 1392, 1270, 1139, 1032 cm⁻¹; for ¹H and ¹³C NMR data, see [Tables 1 and 2;](#page-3-0) ESIMS (positive) m/z 363 [M+Na]⁺; HRESIMS (positive) $[M+Na]^+$ m/z 363.1778 (calcd for $C_{18}H_{28}O_6$ Na, 363.1784).

4.3.19. Compound **19**. Amorphous powder; $\left[\alpha\right]_0^{20}$ -25.0 (c 0.18, GH_CC 0.18, CH_CC 0.18, CH_CC 0.18, CH₂Cl₂); IR (KBr) v_{max} 3523, 3422, 3359, 2934, 2637, 2590, 2370, 1750, 1715, 1660, 1457, 1390, 1076, 945, 912 cm⁻¹; for ¹H and ¹³C NMR data, see [Tables 1 and 2;](#page-3-0) ESIMS (positive) m/z 363 [M+Na]⁺; HRESIMS (positive) $[M+H]^+$ m/z 341.1975 (calcd for C₁₈H₂₉O₆, 341.1964).

4.3.20. Compound **20**. Amorphous powder; $[\alpha]_0^{20}$ +81.1 (c 0.26, MeQH): IR (KBr) π 3523 3090 2925 2869 2361 1727 1654 MeOH); IR (KBr) v_{max} 3523, 3090, 2925, 2869, 2361, 1727, 1654, 1418, 1357, 1319, 1282, 1238, 1164, 1031 cm⁻¹; for ¹H and ¹³C NMR data, see [Tables 1 and 2](#page-3-0); ESIMS (positive) m/z 267 [M+H]⁺; HRE-SIMS (positive) $[M+H]^+$ m/z 267.1594 (calcd for C₁₅H₂₃O₄, 267.1596).

4.3.21. Compound **21.** Amorphous powder; $[\alpha]_0^{20}$ +105.2 (c 0.10, 1660)
MeOH): IR (KBr) 11 - 3552 3239 2942 2567 2361 1739 1660 MeOH); IR (KBr) v_{max} 3552, 3239, 2942, 2567, 2361, 1739, 1660, 1466, 1340, 1266, 1155, 1034, 973, 883 cm⁻¹; for ¹H and ¹³C NMR data, see [Tables 1 and 2](#page-3-0); ESIMS (positive) m/z 267 [M+H]⁺; HRE-SIMS (positive) $[M+H]^+$ m/z 267.1597 (calcd for C₁₅H₂₃O₄, 267.1596).

4.3.22. Compound 22. Amorphous powder; $\left[\alpha\right]_0^{20}$ +54.0 (c 0.16, GH₆Cl₆): IR (KBr) $v = 3439$ 2930 2872 2546 1766 1737 1660 CH₂Cl₂); IR (KBr) v_{max} 3439, 2930, 2872, 2546, 1766, 1737, 1660, 1455, 1370, 1252, 1156, 1031, 998 cm⁻¹; for ¹H and ¹³C NMR data, see [Tables 1 and 2;](#page-3-0) ESIMS (positive) m/z 331 [M+Na]⁺; HRESIMS (positive) $[M+H]^+$ m/z 309.1696 (calcd for C₁₇H₂₅O₅, 309.1702).

4.4. Preparation of (S)-MTPA ester (1a) and (R)-MTPA ester (1b)

Compound 1 (3 mg) was transferred into a clean NMR tube, deuterated pyridine (0.5 mL), small amount of DMAP (dimethyl

amino pyridine), and (R) - $(-)$ - α -methoxy- α -(trifluoromethyl)phe-
pyl acetyl chloride (5 ul.) were added into the NMR tube immedinyl acetyl chloride $(5 \mu L)$ were added into the NMR tube immediately under a N_2 gas stream, and then the NMR tube was shaken carefully to mix the sample and MTPA chloride evenly. The reaction NMR tube was permitted to stand in a water bath at 50 \degree C for 4 h to afford the (S) -MTPA ester derivative $(1a)$. In the manner described for 1a, another portion of compound 1 (3 mg) was reacted in a second NMR tube with $(S)-(+)$ - α -methoxy- α -(trifluoromethyl) phenyl acetyl chloride (5 μ L) at 50 °C for 4 h using deuterated pyridine (0.5 mL) as solvent, small amount of DMAP was added, to afford the (R) -MTPA derivative (1b). The 1 H NMR data of the S-MTPA ester derivative (1a) and R-MTPA ester derivative (1b) were obtained directly on the reaction mixture (pyridine- d_5 , 400 MHz):

4.4.1. Compound 1a. δ 4.458 (1H, dd, J=11.6, 4.4 Hz, H-1), 2.026 (1H, m, H-2a), 1.893 (1H, m, H-2b), 2.625 (1H, m, H-3a), 1.861 (1H, m, H-3b), 1.972 (1H, m, H-4), 2.076 (1H, dd, J=14.0, 11.6 Hz, H-6a), 1.740 (1H, dd, J = 14.0, 7.6 Hz, H-6b), 3.633 (1H, m, H-7), 4.763 (1H, ddd, J = 5.2, 1.2, 1.2 Hz, H-8), 2.588 (2H, d, J=4.8 Hz, H-9), 6.217 (1H, s, H-13a), 5.522 $(1H, s, H-13b), 1.423 (3H, s, H₃-14), 1.015 (3H, d, J=7.6 Hz, H₃-15).$

4.4.2. Compound 1b. δ 4.458 (1H, dd, J=11.6, 4.4 Hz, H-1), 2.007 (1H, m, H-2a), 1.883 (1H, m, H-2b), 2.618 (1H, m, H-3a), 1.857 (1H, m, H-3b), 1.963 (1H, m, H-4), 2.076 (1H, dd, J=14.0, 11.6 Hz, H-6a), 1.740 (1H, dd, J=14.0, 7.6 Hz, H-6b), 3.635 (1H, m, H-7), 4.773 (1H, ddd, J=4.8, 1.2, 1.2 Hz, H-8), 2.600 (2H, d, J=4.8 Hz, H-9), 6.217 $(1H, s, H-13a), 5.522 (1H, s, H-13b), 1.429 (3H, s, H₃-14), 1.015 (3H, d,$ $J=7.6$ Hz, H₃-15).

4.5. Crystallographic data of compound 1 and compound 28

Crystallographic data of compound 1 $C_{15}H_{22}O_4$, $M=266$, tetragonal, space group $P4(1)2(1)2$, $a=8.8042$ (7) Å, $\alpha=90^\circ$; $b=8.8042$ (7) Å, $\beta=90^{\circ}$; c = 39.739 (4) Å, $\gamma=90^{\circ}$; V = 3080.3(5) Å₃ $Z=8$, $D_{\rm{calcd}}=1.226$ mg/m³, crystal size $0.369\times0.344\times0.267$ mm³. Mo K α (0.71073 Å), $F(000) = 1232$, $T = 293(2)$ K. The final R values
were $R = 0.0448$, and R = 0.1195, for 1633 observed reflections were $R=0.0448$, and $R_w=0.1195$, for 1633 observed reflections $[I>2\sigma(I)].$

Crystallographic data of compound 28 $C_{17}H_{24}O_5$, $M=308$, orthorhombic, space group $P2(1)2(1)2(1)$, a=7.9947 (8) Å, $\alpha=90^\circ$; $b=12.3402$ (12) $\rm \AA$, $\rm \AA B=90^\circ$; c=16.8306 (17) $\rm \AA$, $\rm \gamma{=}90^\circ$; V=1660.4(3) $\rm \AA^3$, $Z=4$, $D_{\text{calcd}}=1.234$ mg/m³, crystal size $0.432\times0.320\times0.205$ mm³. Mo $K\alpha$ (0.71073 Å), $F(000)=664$, $T=293(2)$ K. The final R values were
R-0.0413, and R = 0.1034, for 1864 observed reflections $[12\pi/1]$ R=0.0413, and R_w=0.1034, for 1864 observed reflections [$I>2\sigma(I)$].

Crystallographic data for 1 and 28 have been deposited at the Cambridge Crystallographic Data Centre (deposition no. CCDC 776071 and 776072). Copies of these data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: +44 (0)1223 336033 or e-mail: deposit@ccdc.ac.uk).

4.6. In vitro anti-inflammatory assay and cytotoxicity testing

These two experiments were carried out as previously de-scribed.^{[8,9,33](#page-9-0)} Briefly, RAW264.7 cells grown on 100 mm culture dish were harvested and seeded in 96-well plates at 2×10^5 cells/well for NO production. The plates were pretreated with various concentrations of samples for 30 min and then incubated for 24 h with or without $1 \mu g/mL$ of LPS. The nitrite concentration in the culture supernatant was measured by the Griess reaction. Cell viability was measured by an MTT [3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide] assay (Sigma-Aldrich).

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Supplementary data

Supplementary data associated with this article can be found in the online version, at doi:10.1016/j.tet.2010.09.091. These data include MOL files and InChIKeys of the most important compounds described in this article.

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